

2-(2-Fluoro-5-methylbenzoyl)-*N*-phenyl-
2-(1*H*-1,2,4-triazol-1-yl)ethanethioamideLiang-Zhong Xu,* Yong-Qi Qin,
Chong-Yi Zhu, Shuang-Hua Yang
and Kai LiInstitute of Agriculture Chemicals, Qingdao
University of Science and Technology, 266042
Qingdao, People's Republic of China

Correspondence e-mail: qknhs@163169.net

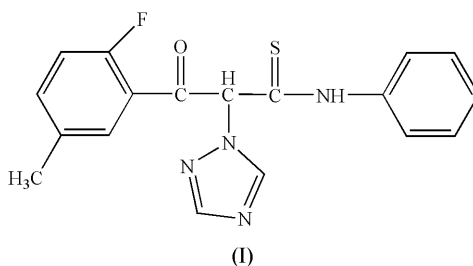
Key indicators

Single-crystal X-ray study
 $T = 293$ K
Mean $\sigma(\text{C}-\text{C}) = 0.005$ Å
 R factor = 0.064
 wR factor = 0.110
Data-to-parameter ratio = 15.5For details of how these key indicators were
automatically derived from the article, see
<http://journals.iucr.org/e>.

The structure of the title compound, $\text{C}_{18}\text{H}_{15}\text{FN}_4\text{OS}$, is stabilized by hydrogen-bonding interactions between the NH group and one of the triazole N atoms of a symmetry-related molecule, resulting in chains parallel to the b axis.

Comment

Recently, compounds containing the 1*H*-1,2,4-triazole group have attracted much interest owing to their fungicidal and plant-growth-regulating activities (Xu *et al.*, 2002), as well as their antibacterial activity against *Puccinia recondite* and root-growth regulation for cucumber (Zhao *et al.*, 1998). In order to search for new triazole compounds with higher bioactivity, we synthesized and characterized the title compound, (I) (Fig. 1).



The bond lengths and angles in (I) agree with those reported in other phenyl and triazole rings (Ji *et al.*, 2002). The C–S bond length is close to the typical C=S bond length. The C–F bond length [1.361 (4) Å] is typical of values found in a related compound with fluorine attached to a benzene ring

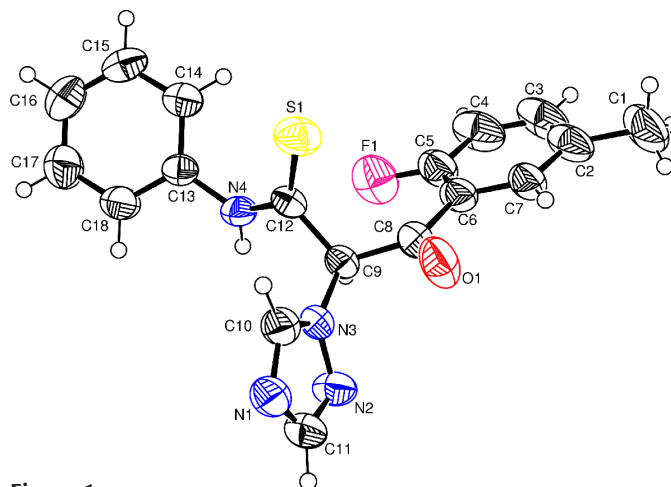


Figure 1

A molecular view (ORTEP-3; Farrugia, 1997) of the title compound, showing 30% probability displacement ellipsoids and the atom-numbering scheme.

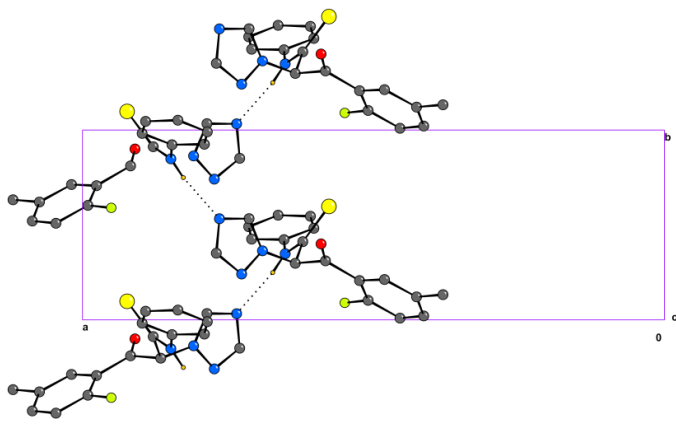


Figure 2
Part of the packing in (I) (CAMERON; Watkin *et al.*, 1993), showing the N–H···N hydrogen-bonding interactions as dashed lines.

[1.363 (7) Å and 1.355 (6) Å in C₁₃H₁₃F₂IO₃ (Mark *et al.*, 2001)]. The carbonyl group is coplanar with the C2–C7 benzene ring (*p*1). The five atoms S1/C9/C12/C13/N4 lie in a plane (*p*2). The dihedral angles formed by the C13–C18 benzene ring and triazole ring with *p*1 and *p*2 are 57.4 (3)/89.8 (6) and 42.1 (3)/83.9 (0)°, respectively.

The most interesting structural feature of (I) is the occurrence of an N–H···N interaction between the NH group and one of the triazole N atoms of a symmetry-related molecule, resulting in the formation of chains parallel to the *b* axis (Table 2 and Fig. 2).

Experimental

The title compound was prepared by reaction of 1-(2-fluoro-5-methylphenyl)-2-(1*H*-1,2,4-triazol-1-yl)ethanone (4.14 g, 0.02 mol), phenyl isothiocyanate (2.24 g, 0.02 mol) and potassium hydroxide (2.24 g, 0.04 mol) in dimethyl sulfoxide solution (30 ml) at room temperature. Single crystals of the title compound suitable for X-ray measurements were obtained by recrystallization from chloroform/ethyl acetate (1:3 *v/v*) at room temperature.

Crystal data

C ₁₈ H ₁₅ FN ₄ OS	<i>D</i> _x = 1.365 Mg m ⁻³
<i>M</i> _r = 354.40	Mo <i>K</i> α radiation
Monoclinic, <i>C</i> 2/ <i>c</i>	Cell parameters from 20 reflections
<i>a</i> = 23.939 (5) Å	<i>θ</i> = 2–11°
<i>b</i> = 7.1430 (14) Å	<i>μ</i> = 0.21 mm ⁻¹
<i>c</i> = 22.029 (4) Å	<i>T</i> = 293 (2) K
<i>β</i> = 113.73 (3)°	Block, yellow
<i>V</i> = 3448.4 (14) Å ³	0.25 × 0.20 × 0.18 mm
<i>Z</i> = 8	

Data collection

Bruker SMART CCD area detector diffractometer	3518 independent reflections
<i>ω</i> scans	1620 reflections with <i>I</i> > 2σ(<i>I</i>)
Absorption correction: multi-scan (SADABS; Sheldrick, 1996)	<i>R</i> _{int} = 0.078
<i>T</i> _{min} = 0.951, <i>T</i> _{max} = 0.963	<i>θ</i> _{max} = 26.4°
9580 measured reflections	<i>h</i> = –29 → 29
	<i>k</i> = –8 → 7
	<i>l</i> = –26 → 27

Refinement

Refinement on <i>F</i> ²	H-atom parameters constrained
<i>R</i> [<i>F</i> ² > 2σ(<i>F</i> ²)] = 0.064	<i>w</i> = 1/[σ ² (<i>F</i> _o ²) + (0.0343 <i>P</i>) ²]
<i>wR</i> (<i>F</i> ²) = 0.110	where <i>P</i> = (<i>F</i> _o ² + 2 <i>F</i> _c ²)/3
<i>S</i> = 1.00	(Δ/ <i>σ</i>) _{max} < 0.001
3518 reflections	Δ <i>ρ</i> _{max} = 0.16 e Å ⁻³
227 parameters	Δ <i>ρ</i> _{min} = –0.14 e Å ⁻³

Table 1

Selected geometric parameters (Å, °).

S1–C12	1.650 (3)	N2–N3	1.362 (3)
F1–C5	1.361 (4)	N3–C10	1.316 (3)
O1–C8	1.203 (3)	N3–C9	1.456 (3)
N1–C10	1.323 (3)	N4–C12	1.325 (3)
N1–C11	1.341 (3)	N4–C13	1.427 (3)
N2–C11	1.311 (3)		
C10–N1–C11	101.8 (3)	N2–N3–C9	119.7 (2)
C11–N2–N3	101.7 (2)	C12–N4–C13	128.7 (2)
C10–N3–N2	109.5 (2)	C12–N4–H4	115.7
C10–N3–C9	130.6 (3)	C13–N4–H4	115.7
C14–C13–N4–C12	44.3 (4)	N2–N3–C9–C8	–87.1 (3)
N2–N3–C9–C12	149.4 (2)	N3–C9–C8–C6	163.4 (2)

Table 2

Hydrogen-bonding geometry (Å, °).

<i>D</i> –H··· <i>A</i>	<i>D</i> –H	H··· <i>A</i>	<i>D</i> ··· <i>A</i>	<i>D</i> –H··· <i>A</i>
N4–H4···N1 ⁱ	0.86	2.07	2.922 (3)	172

Symmetry code: (i) $\frac{1}{2} - x, y - \frac{1}{2}, \frac{1}{2} - z$.

The H atoms were positioned geometrically and were treated as riding on their parent C atoms, with C–H distances in the range 0.93–0.98 Å and *U*_{iso}(H) = 1.2 and 1.5 times *U*_{eq} of the parent atoms.

Data collection: SMART (Siemens, 1996); cell refinement: SAINT (Siemens, 1996); data reduction: SAINT; program(s) used to solve structure: SHELXS97 (Sheldrick, 1997); program(s) used to refine structure: SHELXL97; molecular graphics: ORTEP-3 for Windows (Farrugia, 1997) and CAMERON (Watkin *et al.*, 1993); software used to prepare material for publication: WinGX (Farrugia, 1999).

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